



On Development and Characterisation of a Mobile and Metrologically Traceable Reference Gas Generator for Ammonia and Other Reactive Species in Ambient Air Levels

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MOTIVATION

Ammonia NH₃ in the atmosphere has harmful effects on ecosystems and human health by affecting the acidity of natural waters and soils, thus causing excess nitrogen input [1] and by facilitating secondary particle formation [2]. NH₃ has thus been included in the air quality monitoring networks and emission reduction directives of European nations. Yet, NH₃ data quality, uncertainty estimations and traceability are insufficient [3].

The role of metrology

To enable analyses of trends and effects of reduction measures, reliable and accurate atmospheric measurements are required.

Measurement results have to be traceable via certified NH₃ reference materials (CRM) to be reliable. Well-defined uncertainty contributions [4] and their **uniform assessment ensure the comparability** of measurement results, e.g. of different air quality measuring stations.

The EMRP JRP ENV55 MetNH3 (Metrology for Ammonia in Ambient Air) aims at providing CRM in relevant Ammonia (NH₃) amount fractions (0.5-500 nmol/mol) with expanded (k=2) uncertainty U_{NH3}<3% to the air monitoring community by developing novel approaches for their production [6].

METHOD

Major difficulty: Adsorption of NH₃ on material surfaces causes losses in NH₃ amount fractions of gas mixtures in cylinders

Solution: Dynamic generation of NH₃ CRM

The Federal Institute of Metrology METAS has developed a mobile and metrologically traceable reference gas generator for reactive gases (ReGaS1).

This device is based on the **specific**, **temperature dependent permeation** of the **reference substance (NH₃)** through a membrane into **a flow of carrier gas** (N₂ or air) and subsequent **dynamic dilution** to desired amount fractions in **1 to**

2 steps according to $x_P = x_B \frac{r_1}{r_P + r_P} + x_{DG}$ (where x_P (nmol/mol) is the resulting NH₃ concentration, $x_{\rm B}$ the primary NH₃ concentration, \dot{V}_1 the base gas flow, \dot{V}_2 the dilution gas flow, x_{DG} the background concentration of ammonia in the dilution gas).

ReGaS1 NH3 reference gas mixtures: preliminary uncertainty budget [4]





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Permeation

Continuous, temperature and pressure dependent permeation of pure substance through polymer membrane into Purified matrix gas



Permeation rate $Pr = \frac{\Delta m}{\Delta t}$ of permeation tube is **SI-traceably** determined in temperature and pressure controlled chamber of

Alternatively, permeation tube with calibrated Pr is placed in the **permeation oven** of ReGaS1 flushed by continuous stream of matrix gas.

METAS ReGaS Reactive Gas Standard Mobile, traceable reference gas generator for NH₃



	Rel. expanded measurement uncertainty (f=2)	Rel. contribution to total measurement uncertainty	rer	
Permeation rate	Pr = < 1.7 %	Pr = 20% - 80%	Lor	
Dilution steps (qv _{tot})		$qv_{tot} = < 30 \%$		
1 (qv ₁)	$qv_1 = 0.2 \%$			
Split (qv ₃) $qv_3 = 0.3 \%$			
2 (qv ₄)	$qv_4 = 0.2$ %			
Additional signifi	cant contributions			
Purity of matrix g	as increasing with decrea	sing NH $_3$ amount fractions (0.05 \pm 0	$.05\frac{n}{r}$	
	significantly increases	total measurement uncertainty by up	o to 1	
Temperature reg	ulation Instability increasing	Instability increasing with increasing temperatures, correspo		
in perm. oven	increase in the uncerta	ainty of 2 % for a difference in $T = 0.7$	1 <i>K</i> .	

Related contributions at EGU 2016

• BG2.12 EGU2016-16414; Mo, 18/04/16 15:30-17:00, Foyer M Poster M.25

- AS4.15 EGU2016-8355; Thu, 21/04/16 13:45-14:00, Room -2.47
- AS4.15 EGU2016-11907; Thu, 21/04/16 17:30-19:00, Hall A Poster A.438



ng-term

 $\frac{1mol}{mol}$), 100 % nding to ar



Dilution The traceable NH₃ gas mixture from MSB or Permeation oven is diluted in 1 or 2 traceable steps to atmospheric amount fractions (0.5 - 500 nmol/mol)

First dilution step: Reduces NH₃ amount fractions generated in permeation oven to (> 50 nmol/mol) **Split flow:** Subdivision **Second dilution step:** Reduces NH₃ amount fractions generated in permeation oven to (> 0.5 nmol/mol)

Equal components in MSB and ReGaS1, except for permeation chamber/balance in MSB and permeation oven in ReGaS1: two methods are comparable.



Conclusions

- Quality, uncertainty estimations and traceability of NH₃ data from atmospheric measurement networks are insufficient
- with expanded (k=2) uncertainty $U_{NH3} < 3\%$ to the air monitoring community
- traceable and comparable with the primary reference (MSB).
- commercially available gas generators.
- Generation of NH₃ CRM requires high purity matrix gas:

 - <1.5 nmol/mol
- Reduction of adsorption losses and stabilisation time by surface coating with SilcoNert 2000

References [1] Stevens C J et al.: Nitrogen deposition threatens species richness of grasslands across Europe. Environ. Pollut. 158 2940-5, 2010

[2] Sutton M.A. et al. (2009): Atmospheric Ammonia: Detecting Emission Changes and Environmental Impacts. Springer

[3] EMEP-EEA air pollutant emission inventory guidebook – 2013, Part A, Chapter 5, Table 3-3 [4] ISO-Guide 98-3 "Evaluation of Measurement Data – Guide to the Expression of Uncertainty in Measurement" (GUM).

[5] Vaittinen et al.: Adsorption of Ammonia on stainless steel and treated polymer surfaces. Appl. Phys. B, 2013. [6] Pogány A. et al. : Metrology for ammonia in ambient air - concept and first results of the EMRP project MetNH3, Proc. 17th Intern. Congress of Metrology DO I: 10.1051/metrology/20150 00, 2015



METAS in the framework of EMRP JRP ENV55 MetNH3 (Metrology for Ammonia in Ambient Air) aims at providing CRM in relevant Ammonia (NH_3) amount fractions (0.5-500 nmol/mol)

New development: METAS ReGaS1 mobile reference gas generator as secondary reference

Second dilution step and traceability of all components are novelties compared to

• Ambient air humidity rates (>1% RH) cause dissolution of NH₃ in H₂O multilayers • Matrix gas with $x_{NH3} > 0.05$ nmol/mol cause $U_{NH3} > 3\%$ in NH₃ CRM at amount fractions

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