

On Development and Characterisation of a Portable Standard Gas Generator

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The EMRP JRP ENV55 **MetNH3 (Metrology for Ammonia in Ambient Air)** aims at providing certified reference material (CRM) in relevant Ammonia (NH₃) amount fractions (0.5-500 nmol/mol) for air quality monitoring.

For this purpose and as alternative to CRM in gas cylinders, the Federal Institute of Metrology **METAS is developing a traceable portable gas standard generator**. This device is based on the specific **temperature dependent permeation** of the reference substance (NH₃) through a membrane into a flow of carrier gas (N₂ or air). This mixture is then diluted to desired amount fractions in 1 to 2 steps according to $x_P = x_B \frac{V_1}{V_1 + V_2} + x_{DG}$ (where x_P (nmol/mol) is the resulting NH₃ concentration, x_B the primary NH₃ concentration, \dot{V}_1 the base gas flow, \dot{V}_2 the dilution gas flow, x_{DG} the background concentration of ammonia in the dilution gas).

Second dilution step and traceability of all components are novelties compared to commercially available gas generators.

Here we present **first insights into the development** of said instrument, the **characterisation of individual components** leading to the **uncertainty estimation for the generated NH₃ gas mixture** according to GUM, which is aimed to be **<3 %**.

Moreover, we present **first results of the study on adsorption/desorption effects** of NH₃ on different tubing materials.

Permeation

The controlled permeation of a substance through a polymer membrane is **primarily temperature dependent** and results in a **measurable mass loss over time**.

A permeation tube with a traceable permeation rate $Pr = \frac{\Delta m}{\Delta t}$ (SI-traceably determined by a Magnetic Suspension Balance (MSB), resolution = 10⁻⁶ g) is placed in the permeation oven.

Long-term expanded measurement uncertainty:

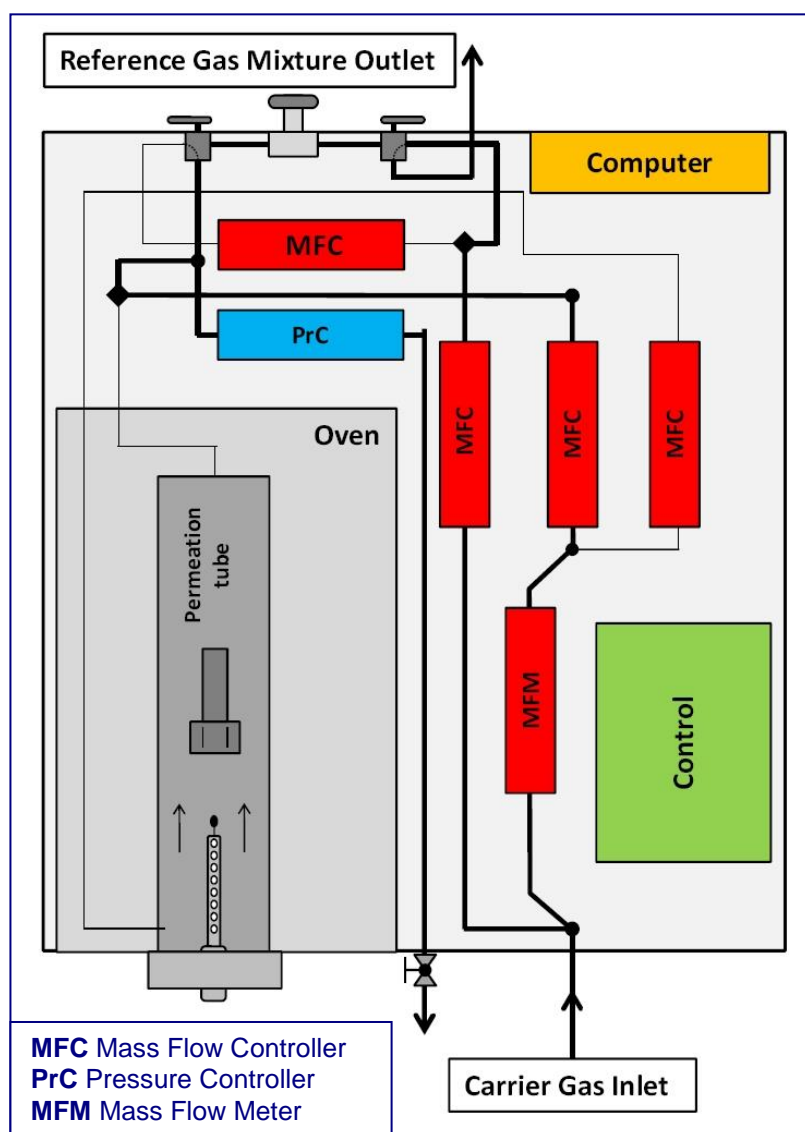
$$Pr = < 1.7 \%$$

Contribution to total measurement uncertainty:

$$Pr = 20\% - 80 \%$$

Surface Coating

Surfaces in contact with NH₃ have been coated with commercially available **SilcoNert 2000** (SilcoTek). This **significantly reduces adsorption losses and stabilisation time** (see below).



Dilution

First dilution step qv_1 : Reduces NH₃ amount fractions generated in permeation oven to (> 100 nmol/mol), expanded measurement uncertainty (k=2) $qv_1 = 0.2 \%$

Split flow qv_3 : Subdivision, expanded measurement uncertainty (k=2) $qv_3 = 0.3 \%$

Second dilution step qv_4 : Reduces NH₃ amount fractions generated in permeation oven to (> 0.5 nmol/mol) expanded measurement uncertainty (k=2) $qv_3 = 0.2 \%$

Contribution of dilution steps to total measurement uncertainty:
 $qv_{tot} = < 30 \%$

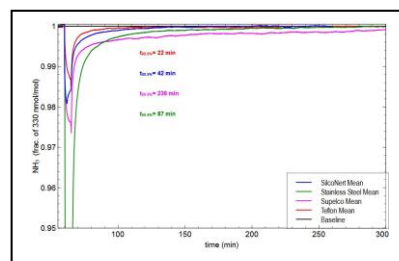
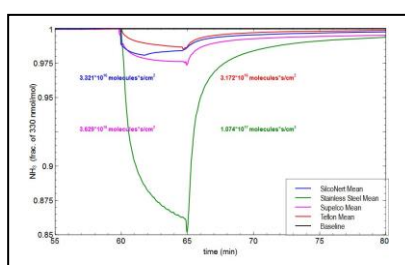
Additional Contributions

Purity of the matrix gas: Increasing with decreasing NH₃ amount fractions (0.05 ± 0.05 nmol/mol), significantly increases total measurement uncertainty by up to 100%.

Temperature regulation in the oven: Instability increasing with increasing temperatures, corresponding to an increase in the uncertainty of 2% for a difference in T = 0.1 K.

Reduction of adsorption losses and stabilisation time by surface coating

Following the approach of **Vaithinen (2013)**, 4 different types of tubing (**Stainless Steel 316**, Polymer (PFA), 2 fused silica coatings applied to Stainless Steel 316 (**Supelco**, **SilcoNert 2000**)) have been exposed to a gas mixture of **330 ppb NH₃ in N₂** (Alphagaz II). **Adsorption loss** (molecules*s/cm²) and **stabilisation time** (t_{99.9%}) have been evaluated.



	t (99.9%)	adsorbed mol*s/cm ² (t _{99.9%})
PFA	22 min	3.172 *10 ¹⁶
SilcoNert 2000	42 min	3.321 *10 ¹⁶
Stainless Steel 316	87 min	1.074 *10 ¹⁷
Supelco	236 min	3.629 *10 ¹⁶

We recommend to have SilcoNert 2000 (SilcoTek) applied to Stainless Steel surfaces in contact with NH₃.