



Dear MetNH3 stakeholder,

EMRP JRP MetNH3 “Metrology for Ammonia in Ambient Air” officially ended on May 31<sup>st</sup> 2017. The project partners would like to thank you very much for your interest and your collaboration over the past 3 years.

More information on MetNH3 can be found, also in the future, on the project webpage <http://www.metnh3.eu>.

Over the course of summer 2017 three “Good Practice Guides” on the following subjects will be made available:

- Good Practice Guide for accurate sampling and preparation of dynamic ammonia gas mixtures
- Good Practice Guide on the measurement system applicable as optical transfer standard for ammonia in ambient air
- Good Practice Guide for measurement of ambient ammonia concentrations in the field and application to policy evidence

The final highlight of the project has been the 2<sup>nd</sup> MetNH3 workshop on ammonia measurement methodology taking place on May 09-10<sup>th</sup> at METAS in Switzerland which was attended by 40 stakeholders. The [book of abstracts](#) is available on the [project website](#).

As one of the main objectives for the participating partners is to develop new capabilities and improved procedures the final newsletter aims at introducing their individual possibilities, particularly those transferable to stakeholders and end-users.

### ***Improved gas mixture standards***

The main objective of WP1 was the **development of improved gas mixture standards for ammonia (NH<sub>3</sub>) by static gravimetric and dynamic methods**. The highly adsorbing/desorbing behaviour of ammonia molecules and the purity of the balance gas challenge the **preparation of ammonia primary reference standards**. The findings of partners NPL, VSL, METAS and BAM will be transferred in the **Good Practice Guide for accurate sampling and preparation of dynamic NH<sub>3</sub> gas mixtures**.

NPL and VSL have tested **different cylinder types** by performing a series of **decant tests at the 100 and 10 μmol mol<sup>-1</sup> levels** and performed **stability tests on those mixtures over a period of up to 24 months**.

VSL performed decanting and stability tests on **3 commercial types of gas cylinders: Spectra Seal (BOC), Aculife 4 (Air Liquide)** and cylinders provided by Takachiho. **All cylinders showed individual**

behaviour in adsorption on the interior surfaces. However, for all tested types of cylinders the stability over long term stability is good. The stability tests performed at VSL show that (Fig. 1):

- The ammonia mixtures of  $100 \mu\text{mol mol}^{-1}$  are stable for 24 months in both Aculife 4 and Spectra Seal cylinders within a 1 % relative uncertainty ( $k = 2$ )
- The ammonia mixtures of  $10 \mu\text{mol mol}^{-1}$  are stable for a period of 12 months within a 3 % uncertainty; cylinders under test were Aculife 4 (AC4), Spectra Seal (SS) and cylinders provided by Takachiho (T)

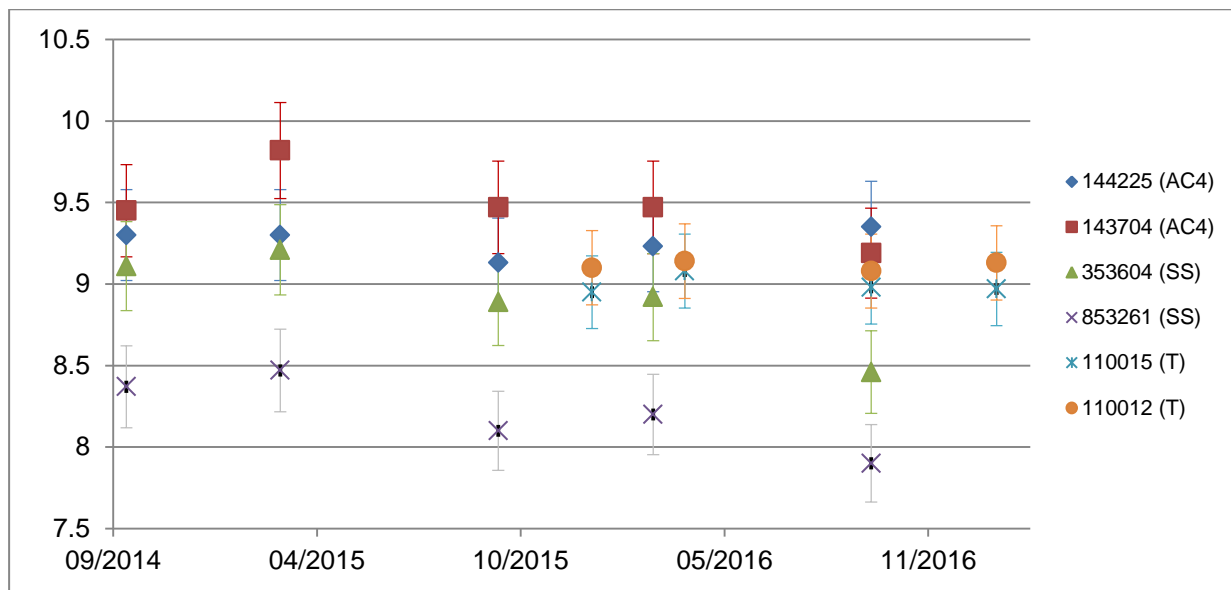


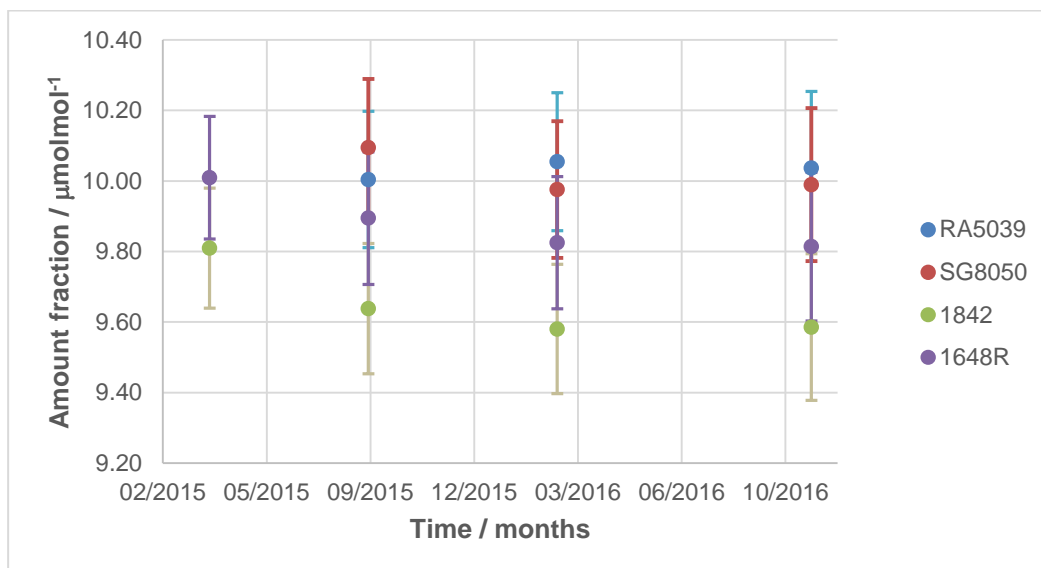
Figure 1: Results for the stability of 6 VSL mixtures at  $10 \mu\text{mol mol}^{-1}$  of  $\text{NH}_3$  in nitrogen. The uncertainty bars represent an expanded uncertainty (with  $k = 2$  for a coverage probability of approximately 95 %).

The stability measurements carried out at NPL show that (Fig. 2):

- the ammonia standard mixtures are completely stable in the BOC cylinders at  $100 \mu\text{mol mol}^{-1}$  level and, after an initial small decay, are stable at  $10 \mu\text{mol mol}^{-1}$ , over the time period studied;
- the SilcoNert2000® treatment applied to stainless steel cylinders is the most suitable for ammonia mixtures at the lowest level of the  $\text{NH}_3$  amount fraction investigated i.e.  $10 \mu\text{mol mol}^{-1}$ . No adsorption loss can be observed.

NPL is experienced in both dynamic and gravimetric preparation of many reference gas mixtures relevant to the atmospheric measurement community. Among the several Calibration and Measurement Capabilities (CMCs) that NPL has, it is worth to mention the CMC for  $\text{NH}_3$  in nitrogen NPL Primary Reference Materials (NPL PRMs) and NPL Calibrated Gas Mixtures (NPL CGMs) in the  $10 - 100 \mu\text{mol mol}^{-1}$  range, with an expanded uncertainty ( $k = 2$ ) of 10 – 5 % respectively. During

the MetNH3 project, 8 NH<sub>3</sub>/N<sub>2</sub> NPL PRMs, in the 10 - 2000 μmol mol<sup>-1</sup> range, and one 140 μmol mol<sup>-1</sup> NH<sub>3</sub>/N<sub>2</sub> NPL CGM have been sold.



**Figure 2:** Results for the stability of 4 NPL mixtures at 10 μmol/mol of NH<sub>3</sub> in nitrogen of which RA5039 and SG8050 are SilcoNert2000 coated cylinders and 1842 and 1648R are BOC cylinders treated with Spectraseal. The uncertainty bars represent an expanded uncertainty (with  $k = 2$  for a coverage probability of approximately 95%).

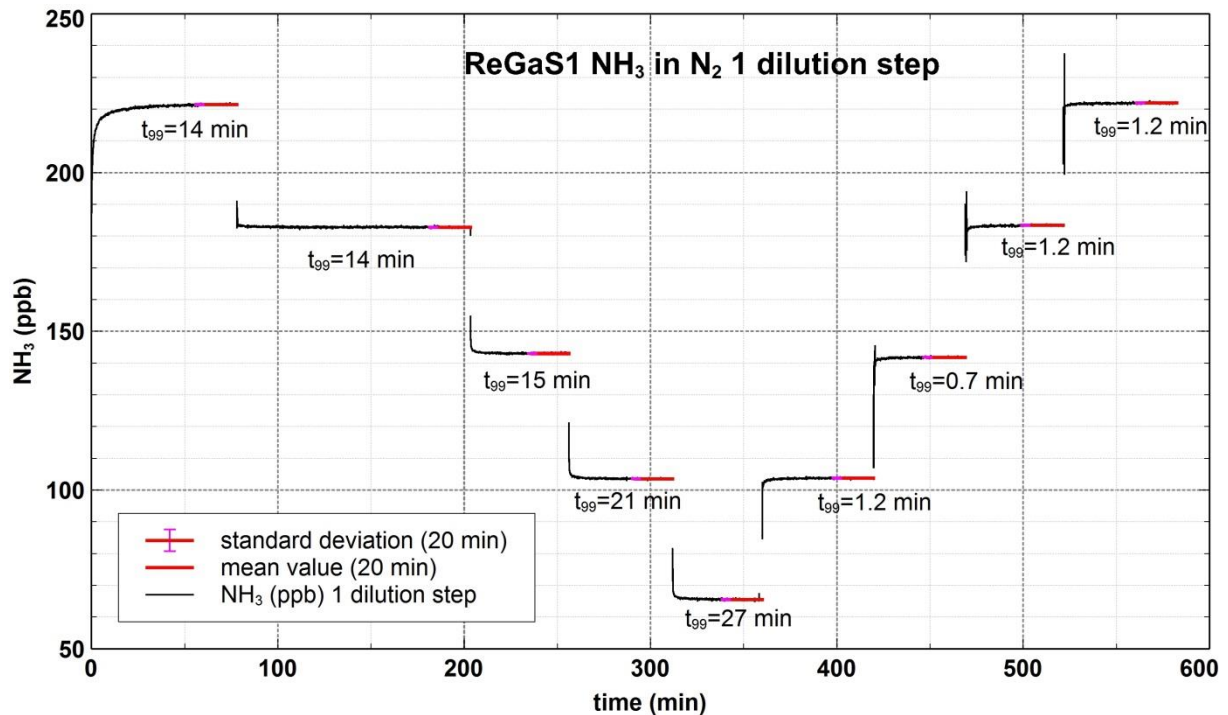
Over the course of MetNH3 **VSL adapted its preparation procedure** to reduce adsorption by minimising the surface exposed to NH<sub>3</sub> during decanting and by applying only coated materials. Moreover, the **analysis and verification procedure for NH<sub>3</sub> mixtures were optimised, thus reducing the analytical uncertainty from 3 % relative to 1 % relative on the 30 μmol mol<sup>-1</sup> amount fraction.** Currently, VSL has CMCs for ammonia in the range from 30 - 300 μmol mol<sup>-1</sup> with an expanded uncertainty of 3 – 2 % relative ( $k=2$ ) and provides calibration services and reference materials world-wide.

VSL is responsible for **organising the next key comparison between national metrology institutes (CCQM-K117) for NH<sub>3</sub> in N<sub>2</sub> gas mixtures** in collaboration with NIST (USA). **The knowledge gained within MetNH3 should enable VSL and other participating institutes (e.g. NPL and METAS) to expand their NH<sub>3</sub> CMCs to a range of 10 – 300 μmol mol<sup>-1</sup> with an expanded uncertainty of 3 – 1 % relative ( $k = 2$ ).**

### ***New dynamic calibration capabilities***

METAS has considerable experience in the dynamic-gravimetric generation of reference gas mixtures combining the permeation method (ISO6145-10) with the dynamic dilution using thermal mass flow controllers (ISO6145-7). The level of uncertainty in the dynamic generation is unaffected

by loss of  $\text{NH}_3$  molecules through adsorption yet, adsorption increases stabilisation times. Thus, **tests of the effects of adsorption and desorption** were carried out on different surfaces of materials widely used for the preparation of reference gas mixtures (e.g. PFA, PTFE, stainless steel). In addition the **influence of humidity on the adsorption of  $\text{NH}_3$**  in the preparation of reference gas mixtures was assessed. It was concluded that adsorption on stainless steel surfaces in contact with  $\text{NH}_3$  can be minimised by up to 90 % when a silica-based coating (SilcoNert2000®, SilcoTek Inc.) is applied on gas-wetted surfaces. Thus, for the **construction of a mobile reference gas generator (ReGaS1) combining permeation and dynamic dilution** SilcoNert2000® has been applied to all surfaces in contact with  $\text{NH}_3$ . **ReGaS1 is a development beyond the state of the art** as it allows for the on-site permeation and dynamic dilution over two dilution steps and thus for the **dynamic generation of  $\text{NH}_3$  at ambient amount fractions (0.5 – 500 nmol mol<sup>-1</sup>) with  $U_{\text{NH}_3} \leq 3 \%$  (Fig. 3)**. ReGaS1 is available for rent for applications in the field and in laboratories (applied in the three MetNH3 intercomparisons). In addition, **METAS offers calibration services** for extractive optical  $\text{NH}_3$  instrumentation as well as for gas mixtures in pressurised cylinders ( $< 300 \mu\text{mol mol}^{-1}$ ), the latter at very low gas consumption from the cylinder to be tested. **The next [NH<sub>3</sub> calibration campaign](#) at METAS will be carried out in autumn 2017.**



**Figure 3:** Different  $\text{NH}_3$  in  $\text{N}_2$  amount fractions generated with ReGaS1 using one dilution step, recorded with a CRDS. The response to a new concentration setting is imminent and the stabilisation time at a new concentration level very fast and with variability  $\leq 0.5 \%$ . The uncertainty in the generated  $\text{NH}_3$  amount fractions is  $\leq 1.5 \%$  for  $\text{NH}_3 > 10 \text{ nmol mol}^{-1}$  and  $\leq 3.0 \%$  for  $\text{NH}_3 > 10 \text{ nmol mol}^{-1}$  (with  $k = 2$  for a coverage probability of 95%).

A similar **mobile gas standard generator (GAS12)** has been developed at **BAM** also combining the permeation method with dynamic dilution to the relevant amount fraction range (0.5 – 500 nmol mol<sup>-1</sup>), yet with slightly higher uncertainty  $U_{\text{NH}_3} \leq 3.7\%$ . It can be applied for the calibration of extractive optical methods and sensor measuring devices.

### ***Improved optical transfer standards***

The goal of **WP2** was the realisation of **optical transfer standards (OTS)** for **NH<sub>3</sub>**, i.e. **IR spectrometric analysers performing absolute NH<sub>3</sub> amount fraction measurements**. Such instrumental standards do not rely on a calibration with NH<sub>3</sub> gas standards but use stable molecular parameters (so-called spectral line parameters) to describe the light absorption properties of ammonia and to deduce the amount fraction in air samples. This **opens a complementary route to traceability and instrument calibration** and circumvents difficulties in the temporal stability of NH<sub>3</sub> reference gas standards normally used for instrument calibration.

An extractive, commercial cavity ring-down (CRDS) instrument by Picarro Inc. with a detection limit < 1 nmol mol<sup>-1</sup> was used by PTB as a starting point to realise an OTS. **PTB developed an OTS in collaboration with DFM by accomplishing the following steps:**

1. **Writing a new validated spectral data evaluation routine** enabling absolute measurements including uncertainty contributions from all relevant parameters in compliance with the GUM
2. **Re-measuring and refining required infrared spectral parameters of ammonia** (line strengths) with unprecedented measurement uncertainty (see publication list)
3. **Metrological characterisation of the instrument**
4. **Performing validation measurements** of reference gas mixtures
5. **Implementing the approach** in a field comparison of NH<sub>3</sub> analysers

This approach will be described in more detail in the **Good Practice Guide on the measurement system applicable as optical transfer standard for ammonia in ambient air**. From the field trial, the expanded uncertainty was found to be approximately  $U(x_{\text{NH}_3}) \approx 2.4\% x_{\text{NH}_3} + 0.8 \text{ nmol mol}^{-1}$  ( $k = 2$ ). Under laboratory conditions with dry gas samples, the performance is better, approximately  $U(x_{\text{NH}_3}) \approx 2.0\% x_{\text{NH}_3} + 0.2 \text{ nmol mol}^{-1}$ . The OTS developed in MetNH<sub>3</sub> extends PTB's portfolio of absolute spectroscopic methods to trace ammonia in air.

### ***State of the art test facility commissioned***

As part of the **planned NH<sub>3</sub> sensor validation programme in WP3**, NPL's **Controlled Atmosphere Test Facility (CATFAC)** was commissioned in the early stages in MetNH<sub>3</sub> and employed for the first systematic exposure testing of ambient ammonia passive samplers. **Seven 28-day exposure tests at different concentrations with ALPHA, Gradko, PASSAM and Radiello diffusive samplers (14 days),**

and DELTA denuders. The exposed samplers were returned to the different manufacturers for analysis. Fig.4 shows the percentage deviation reported by each participant from the known traceable concentrations of NH<sub>3</sub>. Some measurements were in good agreement with the known traceable reference concentrations while other devices exhibited over-reading and under-reading thereby clearly demonstrating the need for this validation work to be carried out.

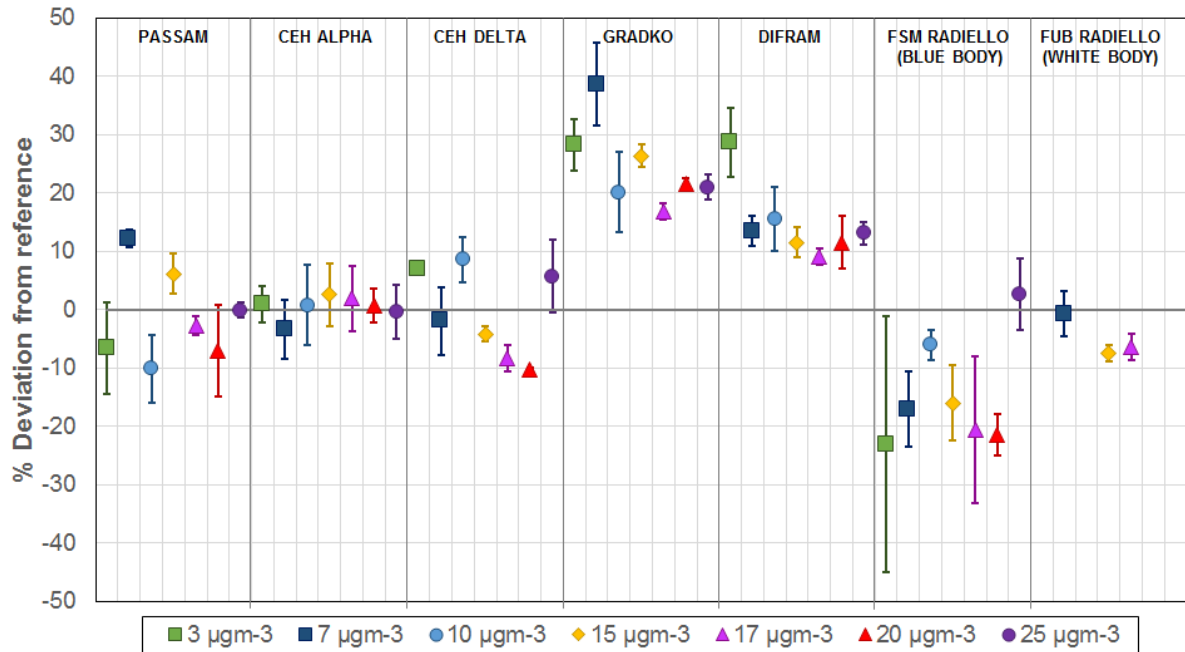


Figure 4: Summary of the mean of the reported NH<sub>3</sub> concentrations for diffusive and pumped samplers tested in the CATFAC, expressed as a percentage deviation from the reference values.

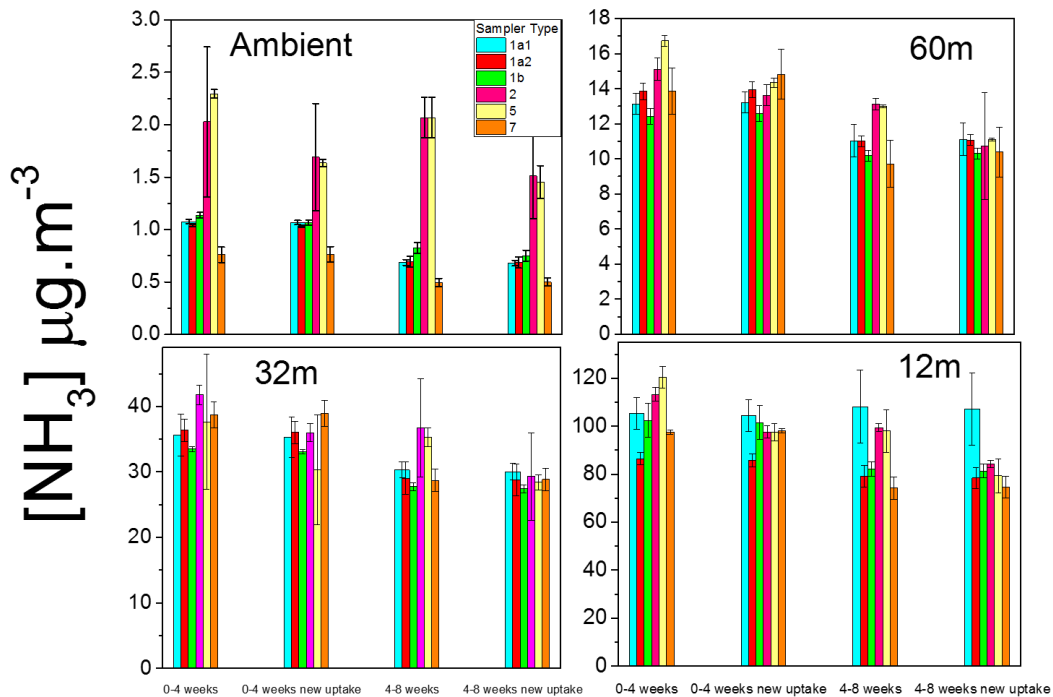
Manufacturer	Diffusive sampler	Sampling rate	Reference T	Sampling rate	Reference T
		U/ m <sup>3</sup> h <sup>-1</sup>	°C	U/ m <sup>3</sup> h <sup>-1</sup>	(°C)
		<b>Measured, this work</b>		<b>Manufacturer provided</b>	
CEH	ALPHA sampler	3.49 x 10 <sup>-3</sup>	20 ± 1	3.45 x 10 <sup>-3</sup>	20
				3.24 x 10 <sup>-3</sup>	10
Gradko	3.5 cm diffusion tube	2.00 x 10 <sup>-4</sup>	20 ± 1	1.62 x 10 <sup>-4</sup>	20
Gradko	DIFRAM-400	3.15 x 10 <sup>-3</sup>	20 ± 1	2.82 x 10 <sup>-3</sup>	20
PASSAM	Passam ammonia sampler	1.88 x 10 <sup>-3</sup>	20 ± 1	1.89 x 10 <sup>-3</sup>	20
Fondazione Salvatore	Radiello sampler (blue body)	1.14 x 10 <sup>-2</sup>	20 ± 1	1.37 x 10 <sup>-2</sup>	20
				1.41 x 10 <sup>-2</sup>	25
Maugeri (FSM)	FUB Radiello sampler (white body)	1.28 x 10 <sup>-2</sup>	20 ± 1	Not Applicable	Not Applicable

Table 1: Summary of NPL CATFAC measured and manufacturer diffusive sampling rates of NH<sub>3</sub> passive samplers.

The results were employed to **determine new provisional values of the diffusive sampling rates for the devices tested**. Table 1 summarises the provisional data obtained. This work has provided the tools to improve the chain of traceability for low concentration ambient measurements of ammonia.

**Two international field measurement intercomparisons hosted**

CEH has more than 20 years' experience in the field application of NH<sub>3</sub> measurements and supported the application of new standards and processes by designing and hosting two intercomparison studies as part of the MetNH3 project: one for intercomparing commercial passive samplers and the second for high resolution active ammonia measurement. Both took place in the summer of 2016 and were open application events. The passive sampler intercomparison was linked to the CATFAC facility experiment (see above). All participants took part in both, with additional participants in the field experiment. The contacts for all the sampler providers including CEH are listed below.

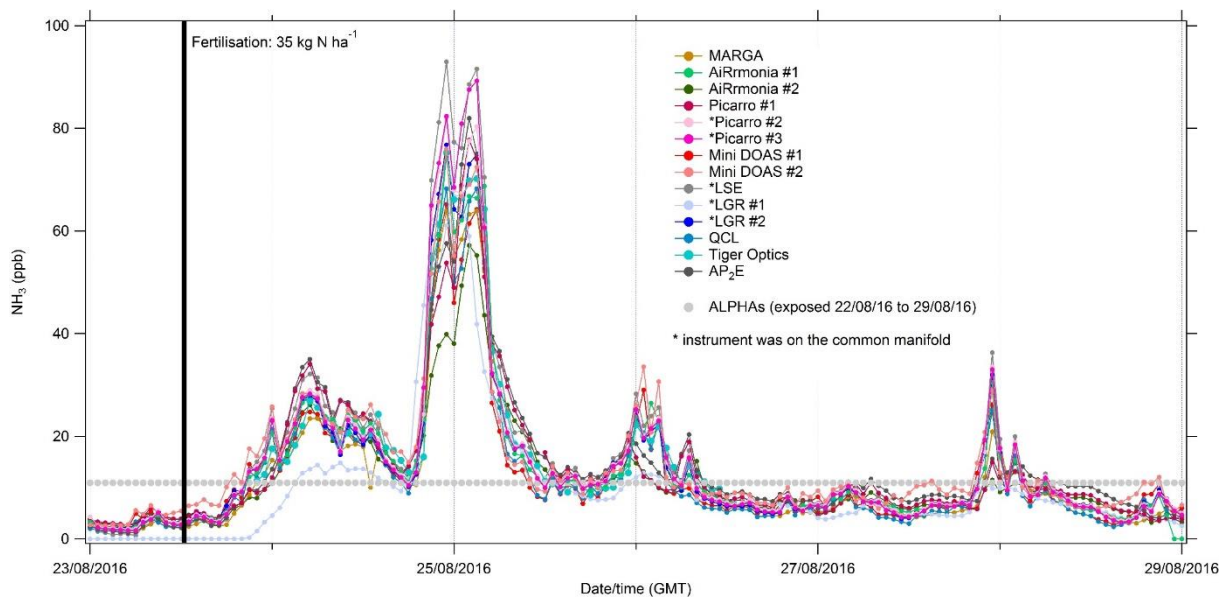


**Figure 5** Diffusion samplers exposed over 4 week periods, with both the manufacturer uptake rate and the new uptake rate that was derived by the CATFAC study in WP3. 1a1 ALPHA CEH Edinburgh 1, 1a2 ALPHA CEH Edinburgh 2, 1 b ALPHA CEH Lancaster, 2 Gradko DIFRAM, 5 Gradko tubes, 7 Passam.

The NH<sub>3</sub> passive sampler field application intercomparison study as part of WP3, was undertaken at an artificial NH<sub>3</sub> release on a peatland site that simulates NH<sub>3</sub> emissions from a medium range



**poultry farm.** Samplers were placed at 12m, 32m and 60m from the emission source, as well as at a background site (ambient concentrations). In total there were 8 participants in the intercomparison with 12 different sets of samplers being tested. **Samplers were exposed for a period of 2 or 4 weeks** dependent on the recommendations by the supplier, over the period from the 16<sup>th</sup> August 2016 to 11<sup>th</sup> October 2016. The results are summarised in Fig. 5. **All samplers and laboratories had good data capture and excellent quality assurance results.** Fig. 5 shows the reported concentrations for two periods for samplers/laboratories which had taken part in the CATFAC study and the impact of the new uptake rates had on the agreement between samplers. **Following this study a peer reviewed publication are currently being written and an annual quality assurance activity based on the intercomparison has been integrated into the UK's National Ammonia Monitoring Network in 2017 and 2018.**



**Figure 6:** Provisional results of the NH<sub>3</sub> intercomparison study held in South East Scotland.

The **NH<sub>3</sub> on-line ambient ammonia measurement field intercomparison study** was held at an intensively managed grassland in south east Scotland from the 21 August 2016 to the 1<sup>st</sup> September 2016. In total **15 participants, including SMEs, regional and national institutions and MetNH3 project partners** took part in the intercomparison and **10 different measurement technologies were applied.** Several techniques were replicated but with different field set-ups, which has allowed an assessment of the effect of set-up on ammonia measurement to be undertaken. Fig. 6 is a provisional summary of the ammonia concentration measurement made over the full intercomparison period by all instruments. It is expected that a peer-review publication will be submitted and the observations from the exercise incorporated into the **Good Practice Guide for measurement of ambient ammonia concentrations in the field and application to policy evidence.**



**Ammonia passive sampler providers and institutes which participated in the field passive sampler intercomparison:**

<p><b>CEH Edinburgh (UK): <a href="#">ALPHA samplers</a></b> Contact name/s: <a href="#">Amy Stephens</a> and <a href="#">Sim Tang</a></p>	<p><b>CEH Lancaster (UK): ALPHA samplers</b> Contact name/s: <a href="#">Jan Poskitt</a></p>
<p><b>IVL (Sweden): IVL samplers</b> Contact name/s: <a href="#">Martin Ferm</a></p>	<p><b>FUB (Switzerland): Radiello and IVL samplers</b> Contact name/s: <a href="#">Eva Seitler</a> and <a href="mailto:fub@fub-ag.ch">fub@fub-ag.ch</a></p>
<p><b>Fondazione Salvatore Maugeri, FSM (Italy): Radiello</b> Contact name/s: <a href="#">Paolo Sacco</a></p>	<p><b>Gradko (UK): diffusion tubes and DIFRAM samplers</b> Contact name/s: <a href="#">Linda Gates</a></p>
<p><b>RIVM (NL): diffusion tubes</b> Contact name/s: <a href="#">Ariën Stolk</a></p>	<p><b>Passam (Switzerland): Passam samplers</b> Contact name/s: <a href="#">Jean-Marc Stoll</a></p>

**Peer-reviewed publications**

The following **peer-reviewed publications** have to date been published in the framework of MetNH3:

- Andrea Pogány et al.: [A metrological approach to improve accuracy and reliability of ammonia measurements in ambient air](#). Meas. Sci. Technol. (2016) **27** 115012
- Nicholas A. Martin et al.: [The application of a cavity ring-down spectrometer to measurements of ambient ammonia using traceable primary standard gas mixtures](#). Appl. Phys. B (2016) 122: 219.
- A. Pogány et al.: [High-Accuracy Ammonia Line Intensity Measurements at 1.5 μm](#). Imaging and Applied Optics 2016, paper JT3A.15.

## Contact and further information

You are very welcome to forward our **newsletter** to your contacts interested in metrology for ammonia in ambient air. **Contact any JRP partner for further information.**

**METAS** (Federal Institute of Metrology), Switzerland. [Bernhard Niederhauser](#) (Coordinator)

**BAM** (Federal Institute for Materials Research and Testing), Germany. [Dr. Carlo Tiebe](#)

**DFM** (National Metrology Institute), Denmark. [Dr. David Balslev-Harder](#)

**MIKES Metrology** (VTT Technical Research Centre), Finland. [Dr. Timo Rajamäki](#)

**NPL** (National Physical Laboratory), United Kingdom. [Dr. Nicholas Martin](#)

**PTB** (National Metrology Institute), Germany. [Dr. Nils Lüttschwager](#)

**UBA** (Federal Environment Agency), Germany. [Dr. Klaus Wirtz](#)

**VSL** (National Metrology Institute), The Netherlands. [Janneke van Wijk](#)

## Researcher Excellence Grants

**REG1: CEH** Centre for Ecology and Hydrology, United Kingdom. [Dr. Christine Braban](#)

**REG2: UH** University of Helsinki, Finland. [Dr. Olavi Vaittinen](#)

The JRP partners have been regularly consulted by an external [board of advisors](#)

